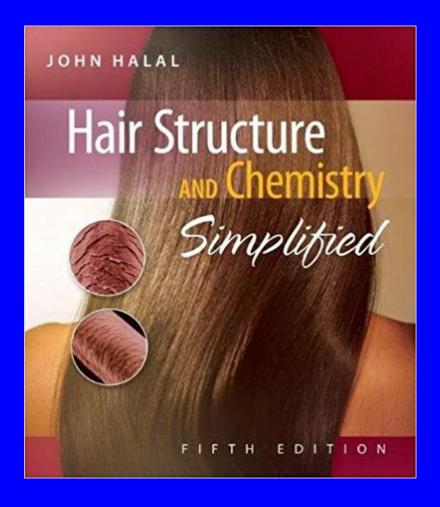
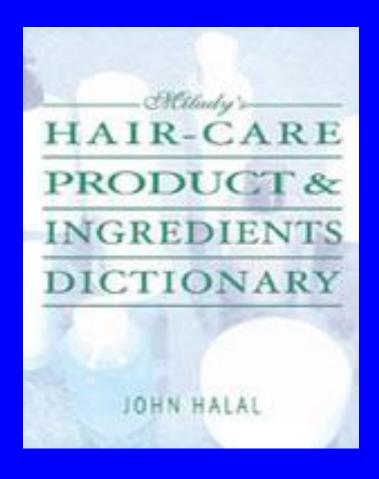


www.chemistrysimplified.com You Tube – How Beauty Works Facebook – John Halal Instagram – johnhalal9

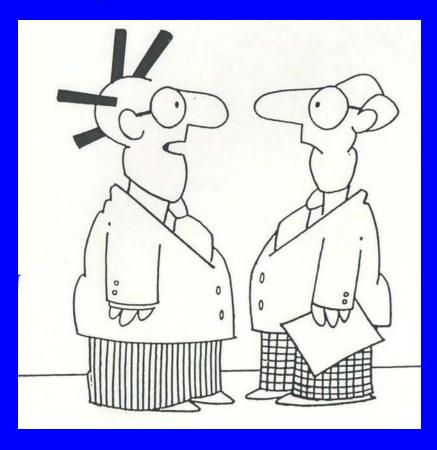


Amazon.com: hair structure and chemistry simplified



<u>Hair-Care Product and Ingredients Dictionary (Milady's Hair Care Product Ingredients Dictionary): Halal, John: 9781562539191: Amazon.com: Books</u>

I don't have much hair left, but my special shampoo fortifies each strand and makes the remaining hair look thicker and fuller.



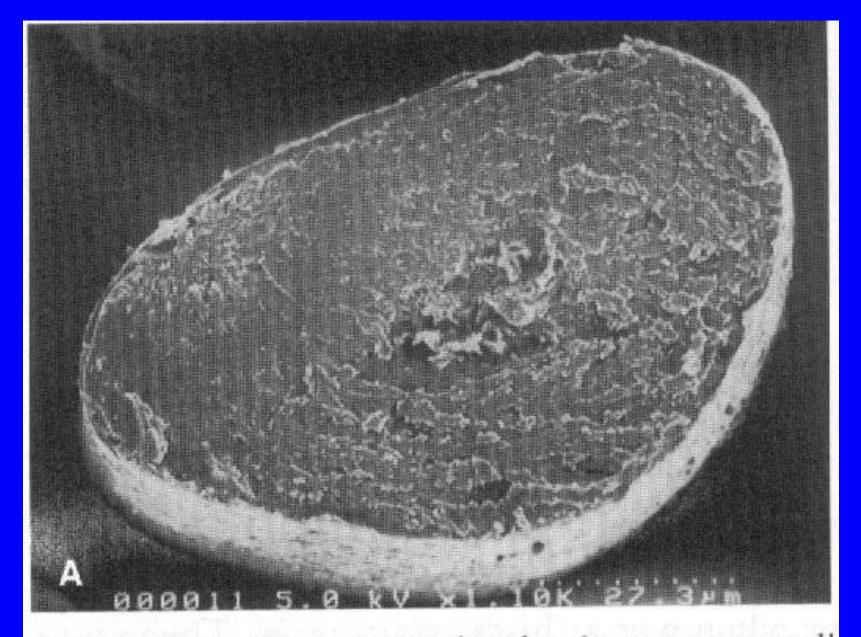
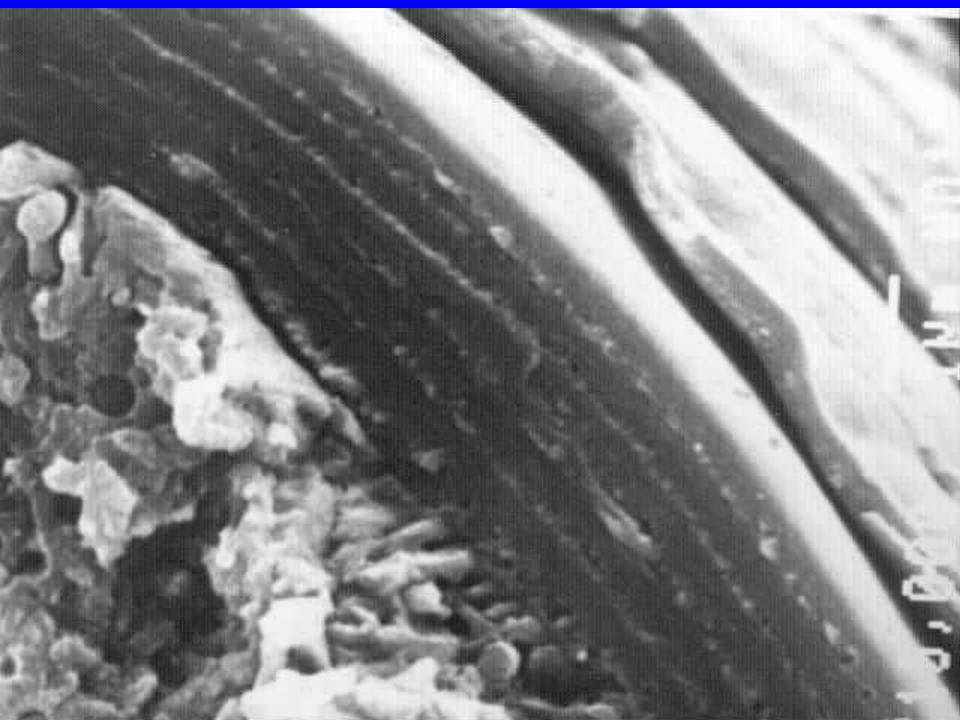


Figure 4. A. Extension to break when wet generall S. Ruetsch, TRI Princeton, B. Extension to break wh



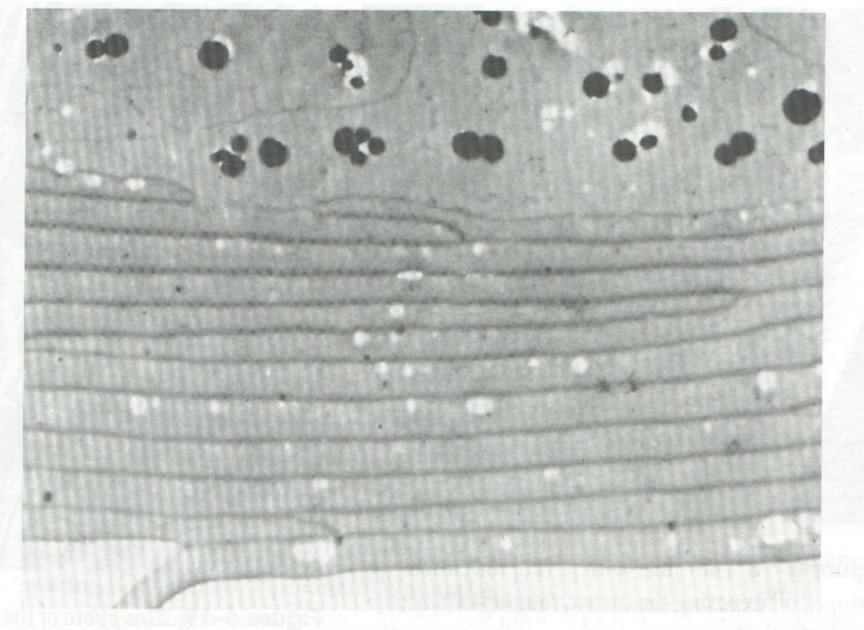
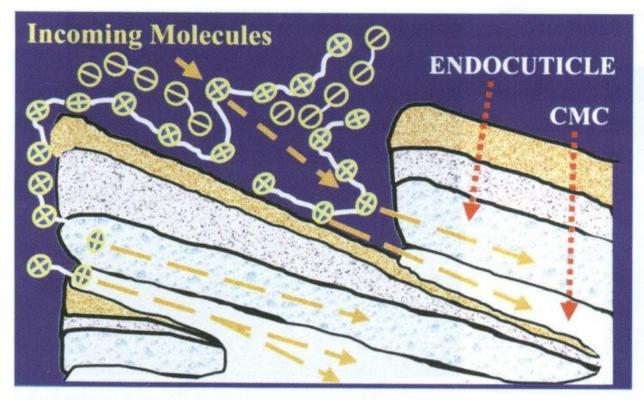


Figure 6-16 A photo of cuticle scales around the cortex, magnified 14,800 times.



Figure 6-15 Cross section of hair showing that, although you can count six distinct layers of overlapping cuticle, each individual scale is attached to the cortex in one cuticle layer.



**Figure 1:** Components of the cell membrane complex and endocuticle are the assumed main diffusion paths of molecules into the hair shaft.

# Penetration Occurs Between The Cuticle Scales





## Subtractive Color Reflects Available Light

What color is your hair
If you turn out all the lights?

- Fluorescent Lights Green
- Tungsten Lights Orange
- Bright Daylight White

	NATURAL LIGHT CONDITIONS	KELVIN COLOR TEMPERATURE	ARTIFICIAL LIGHT SOURCE	COLOR TINT MIXTURE
	• Fair weather, blue sky	- 10,000	Color television	Bluish
	<ul> <li>Slightly cloudy sky</li> <li>Cloudy or rainy sky</li> <li>Sunlight in clear weather at midday</li> <li>Average sunlight in clear weather</li> <li>Sunlight 2 hours after sunrise and before sunset</li> <li>Sunlight 40 min. after sunrise</li> </ul>	-8,000 -7,000 -6,500 -6,000 -5,500 -5,000 -4,500 -4,000 -3,500	<ul> <li>Fluorescent lamp (Daylight)</li> <li>Camera flash bulb</li> <li>Blue lamp for photography</li> <li>Fluorescent lamp (White)</li> <li>Normal flash bulb</li> <li>Fluorescent lamp (Off-white)</li> <li>Tungsten lamp for photography</li> </ul>	Whitish
J.	and before sunset		Halogen lamp     lodine lamp	÷
	<ul> <li>Sunlight 30 min. after sunrise and before sunset</li> <li>Sunlight 20 min. after sunrise and before sunset</li> </ul>	- 2,800	<ul> <li>Tungsten lamp</li> <li>Acetylene lamp</li> <li>Kerosene lamp</li> </ul>	Yellowish
		- 2,000	Candlelight	Reddish

#### **Color Theory**

Shade or Hue of Color
 Balance of Color, Tone or Hue
 Which Color?
 What color is equal parts Yellow, Red and Blue?

# 2) Level of Color Concentration, Density, or Saturation How Much Color? White, Black, & Gray are the Same Color Different Levels Pink is a Lighter Tint of Red



### 10 Stages of Decolorization

- Lightening Natural Color Makes it Warmer
- Natural Hair Color
  - 3 Parts Yellow
  - 2 Parts Red
  - 1 Part Blue

#### Oxidative & Non-oxidative Color

- 1) Non-oxidative
  - Temporary
  - Semi-Permanent
- 2) Oxidative
  - Demi-Permanent (Deposit Only)
  - Permanent (Lift & Deposit)
  - All Lighteners

#### "Coal Tar" Dyes

- Primary Intermediates Provide Base Color
  - Para-Phenylenediamines (PPD)
  - Para-Aminophenols (PAP)
- Couplers/Secondary Intermediates— Modify Color
  - Meta-Phenylenediamines
  - Meta-Aminophenols
  - Resorcinols
- Cosmetic Toiletries & Fragrance Act of 1938
  - Patch Test 48 hours prior to EACH application
  - Not for use on eyelashes or eyebrows

#### Alkaline Haircolor

- 1) Swells the hair to allow penetration.
- 2) Triggers the decomposition of peroxide.
- 3) Aids in developing the dye.

#### Inorganic Alkalizing Agents

- Ammonia, NH<sub>3</sub> (17 aw)
- Water, H<sub>2</sub>O (18 aw)
- Ammonium Hydroxide, NH<sub>4</sub>OH (35 aw)
- Sodium Hydroxide, NaOH (40 aw)

#### Organic Alkalizing Agents

Alkanolamines, R-NH<sub>2</sub>

- Aminomethylpropanol (AMP) (89 aw)
   Deposit Only Over 5 times larger than Ammonia
- Monoethanolamine (MEA) (61 aw)

Lightening – 3.5 times larger than Ammonia

#### Oxidizers

- Oxidation is necessary to:
  - 1) Develop the dye Low Concentration
  - 2) Lighten natural hair color High Concentration
- Examples of oxidizers

Sodium Bromate NaBrO<sub>3</sub> Sodium Perborate NaBO<sub>3</sub>
Hydrogen Peroxide H<sub>2</sub>O<sub>2</sub> Atmospheric Oxygen O<sub>2</sub>
Urea Peroxide CH<sub>2</sub>N<sub>2</sub>O·H<sub>2</sub>O<sub>2</sub>

#### Hydrogen Peroxide

- Peroxide is H<sub>2</sub>O<sub>2</sub>
- Water is H<sub>2</sub>O
- Peroxide is Water with an extra Oxygen
- Peroxide decomposes into Oxygen Gas and Water
  - 1 ounce of 20 Volume produces
  - 1 ounce of water and
  - 20 ounces of Oxygen Gas

### Hydrogen Peroxide H<sub>2</sub>O<sub>2</sub>

- Solution of Hydrogen Peroxide and Water
  - 20 Volume
    - 6% Hydrogen Peroxide
    - 94% Water
  - 40 Volume
    - 12% Hydrogen Peroxide
    - 88% Water

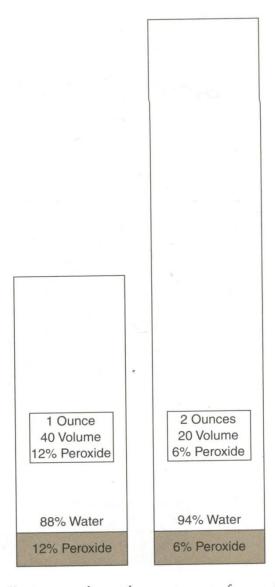


Figure 11-24 This illustration shows that one ounce of 40 volume peroxide contains the same amount of peroxide as two ounces of 20 volume peroxide. The 40 volume simply contains less water.

## When You Mix Equal Parts Color & 20 Volume You Apply 10 Volume to the Hair

#### The Effective Volume is the relationship of:

- 1) The total amount of the color mixture
- 2) To the amount of peroxide
- 3) To the volume of peroxide
- 2 ounces Color
- 2 ounces 20 Volume
- 1) 2+2=4 ounces total
- 2) 4/2 = 2,  $\frac{1}{2}$  is peroxide
- 3) 20/2 = 10 effective volume

- 3 ounces Color
- 1 ounce 40 Volume
- 1) 3 + 1 = 4 ounces total
- 2) 4/1 = 4, 1/4 is peroxide
- 3) 40/4 = 10 effective volume

#### "Resistant" Gray Hair Not Resistant - Just White

- The structure of non-pigmented hair is identical to that of pigmented hair except for the absence of melanin.
- Non-pigmented hair is no more resistant than the pigmented hair on the same head. It's just white.
- "Gray" hair is more difficult to color only because it is white and needs more color.

### Grey Hair Deposit Only

#### Formula

- 3 Ounces Haircolor
- 1 Ounce 30 Volume Peroxide
- 4 Ounces 7.5 Effective Volume
- 3 Parts Yellow, 2 Parts Red, 1 Part Blue
- Level 6 or Darker
- Apply to Clean Damp Hair
- Process 30 minutes at room temperature

#### High Lift

#### Formula

- 1 Ounce Haircolor
- 2 Ounces 40 Volume Peroxide
- 3 Ounces 27 Effective Volume
- Level 8 to 10 with a Blue or Violet Base
- Add ½ ounce of Powder Lightener
- Apply to Clean Damp Hair
- Process 30 minutes, at room temperature

#### Off-The-Scalp Lighteners

- Persulfate Salts
   Ammonium Persulfate, Potassium Persulfate,
   Sodium Persulfate
- Powdered Form Only Anhydrous
   Powder Lighteners, Cream Bleach Activators
- CAUTION SCALP IRRITATION
   Mix the Powder Thoroughly Before Each Use.
   Off Scalp Only
   Do Not Use Heat

#### **Clarifying Treatment**

#### Formula

- 1 Ounce Fruit Fresh (Ascorbic Acid/Vitamin C))
- 1 Ounce Warm Water
- 1 Ounce Clarifying Shampoo with EDTA
- Shampoo with Clarifying Shampoo
- Mix Fruit Fresh and Warm Water
- Add Clarifying Shampoo and Mix Well
- Apply to Damp, Towel Dry Hair
- Cover with Plastic Cap with Holes
- Place Under Hot Dryer for 10 Minutes
- Shampoo with Clarifying Shampoo
- If removing color. Strand test with 10 volume developer.
- If color returns, shampoo again.

# Rate of Reaction Diminishes over time

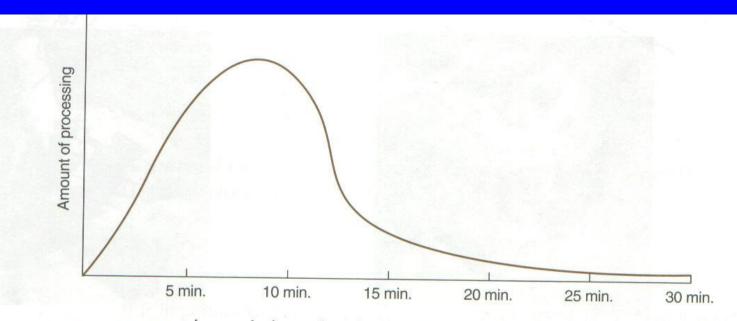
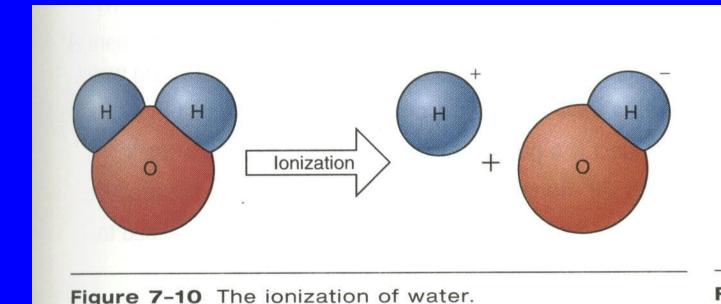


Figure 13-22 The graph shows that the rate of most chemical reactions is not uniform over time. The rate of reaction begins slowly until the chemicals penetrate the hair, then drops sharply as the reactants are "used up" and converted to products. Most chemical reactions have very little chemical activity after 20 minutes.

# pH is due to the lonization of Water H2O → H+ OH-



#### The pH Scale is a Logarithm

- The pH is the negative exponent.
- The pH is the number of decimal places.
- A pH of 7.0 is 50% alkaline & 50% acidic
- pH of 7.0 may be neutral for water, but it's alkaline for hair.
- Place a dollar sign in front of the decimal.
  - Each whole step is a tenfold change.
  - Two whole steps is a one-hundred fold change.

The pH Scale

	H <sup>+</sup> Hydrogen Ion		OH <sup>-</sup> Hydroxide Ion	
рН	Exponential Notation	With Decimal	Exponential Notation	With Decimal
0	1 × 10 <sup>-0</sup>	1.	$1 \times 10^{-14}$	.000000000000001
1	$1 \times 10^{-1}$	.1	$1 \times 10^{-13}$	.000000000001
2	$1 \times 10^{-2}$	.01	$1 \times 10^{-12}$	.00000000001
3	$1 \times 10^{-3}$	.001	$1 \times 10^{-11}$	.00000000001
4	$1 \times 10^{-4}$	.0001	$1 \times 10^{-10}$	.0000000001
5	$1 \times 10^{-5}$	.00001	$1 \times 10^{-9}$	.000000001
6	$1 \times 10^{-6}$	.000001	$1 \times 10^{-8}$	.0000001
7	$1 \times 10^{-7}$	.0000001	$1 \times 10^{-7}$	.0000001
8	$1 \times 10^{-8}$	.0000001	$1 \times 10^{-6}$	.000001
9	$1 \times 10^{-9}$	.00000001	$1 \times 10^{-5}$	.00001
10	$1 \times 10^{-10}$	.000000001	$1 \times 10^{-4}$	.0001
11	$1 \times 10^{-11}$	.0000000001	$1 \times 10^{-3}$	.001
12	$1 \times 10^{-12}$	.00000000001	$1 \times 10^{-2}$	.01
13	$1 \times 10^{-13}$	.000000000001	$1 \times 10^{-1}$	.1
14	$1 \times 10^{-14}$	.00000000000001	1 × 10 <sup>-0</sup>	1.

Figure 9-3 The quantities of the pH scale expressed as pH, exponential notation, and with a decimal point.

## Acid – Alkali Neutralization H<sup>+</sup> + OH<sup>-</sup> → H<sub>2</sub>O

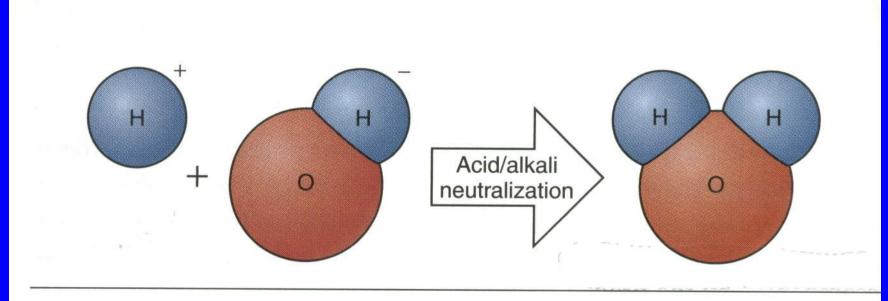


Figure 7-12 Acid/alkali neutralization reaction.

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#### The Copernican Revolution

- The earth is the center of the universe. The sun revolves around the earth.
- Nicolaus Copernicus, 1473 to 1543
- Galileo Galilei, 1564 to 1642
- Sir Isaac Newton, 1642 to 1727
- (1) The Copernican Revolution YouTube